Evaluation of the Physico-chemical Characteristics of Leather Samples of Some Historical Objects from Kiev

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Ukrainian museum and monasteries preserve a large number of parchment and leather documents dated from the 14th to the 19th centuries. As any organic structure, these materials are subjected to the destructive processes due to physico-chemical (light, humidity, temperature, pollutants, etc.), biological and microbiological factors that lead to changes in the polypeptide chain structure. The change resulting from the oxidation, hydrolysis and gelatinization processes in the collagen, can be identified in the IR (middle and NIR), UV-VIS spectra based on some characteristic bands, thermal analysis (DSC) and MHT techniques. These techniques were applied to point out the different kinds of degradation at the microscopic, mesoscopic and molecular levels of the collagen materials.

Keywords: leather, DSC, FTIR and UV-VIS-NIR spectra

Ukrainian museums and monasteries preserve a large number of parchment and leather documents dated from the 14th to the 19th centuries, whose deterioration must be evaluated for selection of the suitable treatment for preservation and restoration.

As any organic structure, these materials are subjected to the destructive processes due to physico-chemical (light, humidity, temperature, pollutants, etc.), biological and microbiological factors that lead to changes in the polypeptide chain structure. The change resulting from the oxidation, hydrolysis and gelatinization processes in the collagen can be identified in the IR (middle and NIR), UV-VIS spectra based on some characteristic bands, thermal analysis (DSC) and shrinkage temperature by MHT method (Micro Hot Table) [1–4].

Based on experience gathered by analyzing the collagen support (leather and parchment) by varied investigation techniques [5–10], our paper presents data regarding the study of leather samples coming from Ukrainian patrimony objects.

Experimental part

Devices
- FT-IR 100 Spectrometer, Perkin-Elmer with vertical ATR, with diamond head;
- UV-VIS-NIR-V670 Spectrometer, Jasco with diffuse reflectance accessory ILN-725; both techniques are non-destructive;
- The DSC curves were recorded using DSC 204 F1 Phoenix apparatus produced by Netzsch – Germany. The following kind of DSC measurements were performed:
  - DSC analysis of the sample (1.12-1.71 mg) in nitrogen flow (purity of gas is higher than 99.999%; 20 mL.min⁻¹), in an open aluminium pan, at the heating rate of 10 K.min⁻¹, and in the temperature range 25°C ... 280°C;
  - DSC analysis of the sample (0.93-2.50 mg) immersed in water (35μL deionised water), hermetically sealed in an aluminium pan and stocked for 24 h. Each sample was heated from 25°C to 110°C, at a heating rate of 10 K.min⁻¹;
- MHT with Caloric equipment (Stereomicroscope Leica), hot table, webcam, computer and software to acquire the image correlated with the shrinkage temperature.

Materials
- P1 book leather, Lvov;
- P2 book leather, Lvov;
- P3 archaeological leather (XV century);
- P4 carriage leather – Historical Museum (Kiev).
- P5 calf leather tanned with quebracho (ICPI, Bucharest)

Results and discussions

Spectral characteristics

In order to identify the degradation types, we selected some IR spectral characteristics:

<table>
<thead>
<tr>
<th>Band (cm⁻¹)</th>
<th>Meaning</th>
</tr>
</thead>
<tbody>
<tr>
<td>3450 – 3200</td>
<td>νOH + νNH</td>
</tr>
<tr>
<td>1660 – 1630</td>
<td>νC=O (amide I)</td>
</tr>
<tr>
<td>1550 – 1530</td>
<td>δNH (amide II)</td>
</tr>
<tr>
<td>1750 – 1700</td>
<td>νC=O (oxygenated structures)</td>
</tr>
</tbody>
</table>

We have computed:
- Δν = ν₁ - ν₂, for information on the denaturation of the polypeptide chain [4, 9, 11].

The main spectral characteristics of the investigated samples are presented in table 1.

In comparison with the standard – calf leather tanned with quebracho – all the historical leathers present a 10–30% higher degree of hydrolysis and oxidation groups coming from the tanning agent (at P5) as well as from oxidative processes taking place in time (P1 - P4) proved by the presence of the free acids (ν 1730 cm⁻¹) and the ketones (ν 1750 cm⁻¹) (fig. 1) [8].

For the UV-VIS-NIR domain, the main bands are:
- bands from the 250 – 500 nm domain, attributed to the π→π * and n→π * like the extended conjugation structure from the tannants;
- the 1470 – 1510 nm band given by the valence harmonic vibrations of the hydroxyl groups ($\nu$OH) from the water absorbed into the collagen structure and which allows the identification of the types and the strength of the hydrogen bond from the peptide chain;
- the 1900 – 1920 nm bands are given by the valence and deformation vibrations ($\nu$OH + $\delta$OH) of the hydroxyl groups from the peptide chain.

In the UV-VIS domain, all the samples present a wide band at 250 – 500 nm due to the – CONH – group and the tanning agent color. In the NIR domain, the bands from 1480 – 1500 and 1900 – 1920 nm point to the presence of the hydroxyl structures associated inter/intra molecular by hydrogen bonds (fig. 2) [12, 13].

The absorbance of the 1480 – 1500 nm band is partially correlated with the shrinkage temperature (ST) pointing to changes in the hydrogen bonds structure.

The main chromatic characteristics are the luminosity ($L^*$) and chroma ($C^*$), while for the differentiation of the samples, we selected the value $\Delta L^* = L_{st}^* - L_s^*$ (st – standard, s – sample).

The change in luminosity varies between 21 – 35%, suggesting an increase of the luminosity but the nuance angle remains in the same domain (50 – 70%).

![Fig. 1. FT-IR spectra of the samples](http://www.revistadechimie.ro)
Thermal characteristics
DSC analysis in \( N_2 \) flow

Figure 3 shows the DSC curves obtained by analysis in \( N_2 \) flow for the investigated samples; similar DSC curves were obtained for many new and old collagen-based materials [5,14]. At a relative low temperature, each investigated material has exhibited an endothermic peak, denoted by I, corresponding to the loss of material humidity. For the samples P1, P2 and P5, this process is followed by one or two endothermic processes, denoted by II. This process, earlier detected by Okamoto and Saeki [15, 16], and recently [5, 14] put in evidence for pure collagen and collagen-based materials, could be explained by the biphasic amorphous-crystalline structure of collagen-based materials according to which the crystalline triple-helix is embedded into an amorphous matrix [5]. Very recently [14], the data from Proton solid-state NMR, obtained for pure collagen, parchments and leathers, have lead to a three-phase model (rigid, inter-face, and mobile phase). Consequently, the process II might be related to softening (melting) of the rigid (crystalline) part of parchment.

The characteristic parameters put in evidence in the DSC curves – \( N_2 \) flow are listed in table 2. According to the curves shown in figure 4 the data listed in table 2:

(a) the standard sample exhibit a single melting process having the parameters in the ranges previously reported [14] for many new leathers manufactured by vegetable tanning \((240^\circ C \leq T_m \leq 252^\circ C; 0.9 \text{ J.g}^{-1} \leq -\Delta H \leq 2.2 \text{ J.g}^{-1})\);

(b) the samples P1 and P2 exhibit melting points in the temperature range 113\(^\circ C \ldots 121\(^\circ C\);

(c) the sample P1 exhibits two melting processes;

(d) in the temperature range in which the analyses were performed, the sample P4 has not a melting process.

The statement (b) could be explained by the results obtained in our previous DSC study of the thermal behavior of leathers and parchments [14]. It was observed that at the heating in both \( O_2 \) and synthetic air flows, all investigated new leathers manufactured by vegetable tanning and old leathers exhibit a softening process at lower temperatures \((118^\circ C \ldots 135^\circ C)\). This means that a thermo-oxidation of new and old leathers occurs when these materials are heated in oxidative atmosphere. It was also
Table 2
THE CHARACTERISTIC PARAMETERS OF THE PROCESSES PUT IN EVIDENCE IN DSC CURVES

<table>
<thead>
<tr>
<th>Sample code</th>
<th>T&lt;sub&gt;m&lt;/sub&gt;°C</th>
<th>-ΔH J.g&lt;sup&gt;-1&lt;/sup&gt;</th>
<th>T&lt;sub&gt;m&lt;/sub&gt;°C</th>
<th>-ΔH J.g&lt;sup&gt;-1&lt;/sup&gt;</th>
<th>N</th>
<th>Process</th>
<th>T&lt;sub&gt;onset&lt;/sub&gt;°C</th>
<th>T&lt;sub&gt;m&lt;/sub&gt;°C</th>
<th>-ΔH J.g&lt;sup&gt;-1&lt;/sup&gt;</th>
</tr>
</thead>
<tbody>
<tr>
<td>P1</td>
<td>49.6</td>
<td>196.3</td>
<td>113.4</td>
<td>1.00</td>
<td>5</td>
<td>I</td>
<td>35.2</td>
<td>38.5</td>
<td>2.83</td>
</tr>
<tr>
<td></td>
<td></td>
<td></td>
<td>128.1</td>
<td>0.87</td>
<td></td>
<td>II+II'</td>
<td>55.1</td>
<td>59.5; 61.7</td>
<td>8.02</td>
</tr>
<tr>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td>III+III'</td>
<td>67.0</td>
<td>68.8; 71.3</td>
<td>1.15</td>
</tr>
<tr>
<td>P2</td>
<td>44.1</td>
<td>184.7</td>
<td>120.8</td>
<td>2.01</td>
<td>4</td>
<td>I</td>
<td>34.6</td>
<td>38.3</td>
<td>0.63</td>
</tr>
<tr>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td>II</td>
<td>46.6</td>
<td>47.9</td>
<td>0.58</td>
</tr>
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<td></td>
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<td></td>
<td>II</td>
<td>54.3</td>
<td>56.8</td>
<td>0.62</td>
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<tr>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td>IV</td>
<td>61.4</td>
<td>65.4</td>
<td>5.17</td>
</tr>
<tr>
<td>P3</td>
<td>52.4</td>
<td>211.5</td>
<td>-</td>
<td>-</td>
<td>4</td>
<td>I</td>
<td>34.3</td>
<td>38.1</td>
<td>0.18</td>
</tr>
<tr>
<td></td>
<td></td>
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<td></td>
<td></td>
<td>II</td>
<td>43.8</td>
<td>46.9</td>
<td>0.12</td>
</tr>
<tr>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td>III+III'</td>
<td>51.1</td>
<td>57.4; 61.7</td>
<td>12.54</td>
</tr>
<tr>
<td>Standard sample P5</td>
<td>67.3</td>
<td>218.8</td>
<td>249.4</td>
<td>1.91</td>
<td>1</td>
<td>I</td>
<td>72.7</td>
<td>78.0</td>
<td>27.61</td>
</tr>
</tbody>
</table>

T<sub>m</sub> = temperature corresponding to the peak minimum; ΔH = process enthalpy; N = number of processes put in evidence in the temperature range 30°C…95°C; X' is the process overlapped with process X.

Table 2 shows the characteristic parameters put in evidence in the DSC curves – water are listed in table 2.

It was pointed out by several authors [16] that the value of the shrinkage temperature determined by classical or MHT methods for a given collagen based material is closed to the extrapolated onset temperature (T<sub>onset</sub> in table 2) corresponding to the main process of denaturation put in evidence in DSC curve recorded for the sample immersed in water. The existence of many peaks in the DSC curve.

Figure 4 shows the DSC curves obtained by analysis of the investigated sample immersed in water.

The DSC analysis of the samples immersed in waters show that in the temperature range 20°C...95°C only the recent manufactured leather exhibit a single endothermic peak. The characteristic parameters put in evidence in the DSC curves – water are listed in table 2.

It was pointed out by several authors [16] that the value of the shrinkage temperature determined by classical or MHT methods for a given collagen based material is closed to the extrapolated onset temperature (T<sub>onset</sub> in table 2) corresponding to the main process of denaturation put in evidence in DSC curve recorded for the sample immersed in water. The existence of many peaks in the DSC curve.
(fig. 5 and table 2) indicates a high degree of material heterogeneity. On the other hand, the DSC investigation of many collagen based materials [5] leads to the following values of $T_{\text{onset}}$: 77.1±8.0°C for recent leathers manufactured by vegetal tanning, 67.0±12.0°C for middle age leathers, 58.0±7.0°C for new and old (middle age) parchments, and 40.1±2.2°C for fresh collagen extracted from animal skins. The data listed in table 2 allow making some similarities between the hydrothermal stabilities of the investigated leathers and the above mentioned collagen based materials. The low values of $T_{\text{onset}}$ and/or the existence of many DSC peaks indicate a high degree of deterioration of all investigated leathers.

Shrinkage temperature

By using the MHT method (Micro Hot Table) the results obtained are shown in the table 3.

The samples P1 – P4 have 30 % lower shrinkage temperature in comparison with the standard (P5).

Conclusion

The samples from Ukraine present a moderate degree of degradation, 20 – 35% losses in the initial luminosity and, on average, a 23% diminishing in the contraction temperature. The shrinkage temperature is semi-quantitatively correlated with the absorbance of the 1480 – 1500 nm bands, pointing to changes in the hydroxyl structures associated by hydrogen bonds.

The DSC analyses of some patrimony leathers were performed in the following conditions: samples immersed in water, hermetically sealed in an aluminum pan and stocked for 24 h; samples in open crucible, under N2 flow. The DSC plots recorded in gas flow show two different processes occurring, namely, the loss of water (dehydration) and the softening (melting). Focusing on the second process observed when the analysis is performed in N2 flow, it was pointed out that unlike the new leathers, and the naturally aged ones exhibit an advanced degree of oxidation.

The results obtained by DSC analyses of sample immersed in water show a high heterogeneity of the old leathers, which could be assigned as a mixture of untanned collagen and sorts of leathers with different degree of cross-linking.

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